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## Structure Reports

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## (2Z)-2-Methylsulfanyl-1,4-di-2-thienyl-but-2-ene-1,4-dione

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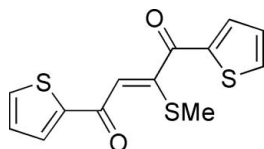
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Key indicators: single-crystal X-ray study;  $T = 297$  K; mean  $\sigma(\text{C}-\text{C}) = 0.003$  Å;  $R$  factor = 0.045;  $wR$  factor = 0.129; data-to-parameter ratio = 17.9.

The asymmetric unit of the title compound,  $\text{C}_{13}\text{H}_{10}\text{O}_2\text{S}_3$ , contains two independent molecules, which exhibit different conformations [ $\text{C}=\text{C}(\text{SMe})-\text{C}(\text{O})-\text{C}$  torsion angles of  $81.4$  (3) and  $80.2$  (2)°], normal bond lengths and angles, and short intramolecular  $\text{S}\cdots\text{O}$  distances [2.768 (3) Å in both molecules]. The crystal packing exhibits weak intermolecular  $\text{C}-\text{H}\cdots\text{O}$  hydrogen bonds.

### Related literature

For details of the preparation of the title compound, see: Yin *et al.* (2006, 2007). For the crystal structures of similar compounds, see: Chen *et al.* (2007); Gao *et al.* (2007). For applications of the title compound, see: Ellison *et al.* (1973); Yuguchi *et al.* (2004).



### Experimental

#### Crystal data

$\text{C}_{13}\text{H}_{10}\text{O}_2\text{S}_3$   
 $M_r = 294.41$   
Monoclinic,  $P2_1/c$   
 $a = 9.7994$  (14) Å  
 $b = 27.571$  (4) Å  
 $c = 10.8814$  (16) Å  
 $\beta = 113.816$  (2)°

$V = 2689.6$  (7) Å<sup>3</sup>  
 $Z = 8$   
Mo  $K\alpha$  radiation  
 $\mu = 0.54$  mm<sup>-1</sup>  
 $T = 297$  (2) K  
0.20 × 0.20 × 0.10 mm

#### Data collection

Bruker SMART 4K CCD area-detector diffractometer  
Absorption correction: multi-scan (SADABS; Sheldrick, 2001)  
 $T_{\min} = 0.900$ ,  $T_{\max} = 0.948$

29719 measured reflections  
5868 independent reflections  
5087 reflections with  $I > 2\sigma(I)$   
 $R_{\text{int}} = 0.022$

#### Refinement

$R[F^2 > 2\sigma(F^2)] = 0.045$   
 $wR(F^2) = 0.129$   
 $S = 1.06$   
5868 reflections

327 parameters  
H-atom parameters constrained  
 $\Delta\rho_{\max} = 0.41$  e Å<sup>-3</sup>  
 $\Delta\rho_{\min} = -0.25$  e Å<sup>-3</sup>

Table 1

Hydrogen-bond geometry (Å, °).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$\text{C}2-\text{H}2\cdots\text{O}4^i$	0.93	2.52	3.448 (3)	172
$\text{C}5-\text{H}5\cdots\text{O}4^i$	0.93	2.54	3.416 (3)	157
$\text{C}20-\text{H}20\cdots\text{O}4^i$	0.93	2.54	3.438 (3)	162

Symmetry codes: (i)  $x, y, z + 1$ ; (ii)  $x + 1, y, z + 1$ .

Data collection: SMART (Bruker, 2001); cell refinement: SAINT (Bruker, 2001); data reduction: SAINT; program(s) used to solve structure: SHELXS97 (Sheldrick, 1997); program(s) used to refine structure: SHELXL97 (Sheldrick, 1997); molecular graphics: PLATON (Spek, 2003); software used to prepare material for publication: PLATON.

The authors are grateful to Central China Normal University and Dr Guo-Dong Yin.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: CV2356).

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**supplementary materials**

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## (2Z)-2-Methylsulfanyl-1,4-di-2-thienylbut-2-ene-1,4-dione

A.-H. Chen and J. Wu

### Comment

1,4-Dicarbonyl compounds are widely used as synthetic building blocks for further elaboration into substituted cyclopentenones, such as jasmones, rethorolones, cuparenones, prostaglandins, and five-membered heterocyclic compounds, such as furans, pyrroles, thiophenes, pyridazines (Ellison *et al.*, 1973; Yuguchi *et al.*, 2004). As a continuation of our previous studies in this area, (Chen *et al.*, 2007; Gao *et al.*, 2007; Yin *et al.*, 2007). We herein report the structure of the title compound (I) (Fig. 1).

There are two independent molecules in the asymmetric unit. The bond lengths of the C1—C2 and C14—C15 in two molecules are 1.339 (2) and 1.341 (2) Å, respectively, supporting their double-bond character. The torsion angles C2—C1—C8—C9 and C15—C14—C21—C22 are  $-81.4$  (3) and  $80.2$  (2) °, respectively, showing the difference in conformations of two molecules. In both molecules, there is a short intramolecular S···O distance of 2.768 (3) Å.

The crystal packing exhibits weak intermolecular C—H···O hydrogen bonds (Table 1).

### Experimental

The title compound was prepared according to the method described by Yin *et al.* (2006). Crystals appropriate for data collection were obtained by slow evaporation from a CH<sub>2</sub>Cl<sub>2</sub>/CH<sub>3</sub>OH (1:1, v/v) solution at 283 K.

### Refinement

All H atoms were geometrically positioned (C—H 0.93–0.96 Å) and refined as riding, with  $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$  or  $1.5U_{\text{eq}}$  (methyl C).

### Figures

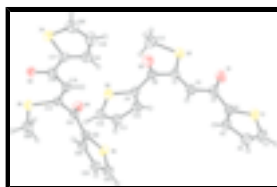


Fig. 1. The content of asymmetric unit of (I) showing the atom-labelling scheme and displacement ellipsoids drawn at the 30% probability level.

## (2Z)-2-Methylsulfanyl-1,4-di-2-thienylbut-2-ene-1,4-dione

*Crystal data*

C<sub>13</sub>H<sub>10</sub>O<sub>2</sub>S<sub>3</sub>

$F_{000} = 1216$

# supplementary materials

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$M_r = 294.41$

Monoclinic,  $P2_1/c$

Hall symbol: -P2ybc

$a = 9.7994$  (14) Å

$b = 27.571$  (4) Å

$c = 10.8814$  (16) Å

$\beta = 113.816$  (2)°

$V = 2689.6$  (7) Å<sup>3</sup>

$Z = 8$

$D_x = 1.454$  Mg m<sup>-3</sup>

Mo  $K\alpha$  radiation

$\lambda = 0.71073$  Å

Cell parameters from 6482 reflections

$\theta = 2.3$ – $28.3$ °

$\mu = 0.54$  mm<sup>-1</sup>

$T = 297$  (2) K

Block, colourless

$0.20 \times 0.20 \times 0.10$  mm

## Data collection

Bruker SMART 4K CCD area-detector diffractometer

Radiation source: fine-focus sealed tube

Monochromator: graphite

$T = 297$ (2) K

$\varphi$  and  $\omega$  scans

Absorption correction: multi-scan (SADABS; Sheldrick, 2001)

$T_{\min} = 0.900$ ,  $T_{\max} = 0.948$

29719 measured reflections

5868 independent reflections

5087 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.022$

$\theta_{\text{max}} = 27.0$ °

$\theta_{\text{min}} = 1.5$ °

$h = -12 \rightarrow 12$

$k = -35 \rightarrow 32$

$l = -12 \rightarrow 13$

## Refinement

Refinement on  $F^2$

Least-squares matrix: full

$R[F^2 > 2\sigma(F^2)] = 0.045$

$wR(F^2) = 0.129$

$S = 1.06$

5868 reflections

327 parameters

Primary atom site location: structure-invariant direct methods

Secondary atom site location: difference Fourier map

Hydrogen site location: inferred from neighbouring sites

H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0723P)^2 + 0.9173P]$$

where  $P = (F_o^2 + 2F_c^2)/3$

$(\Delta/\sigma)_{\text{max}} = 0.001$

$\Delta\rho_{\text{max}} = 0.41$  e Å<sup>-3</sup>

$\Delta\rho_{\text{min}} = -0.25$  e Å<sup>-3</sup>

Extinction correction: none

## Special details

**Geometry.** All e.s.d.'s (except the e.s.d. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell e.s.d.'s are taken into account individually in the estimation of e.s.d.'s in distances, angles and torsion angles; correlations between e.s.d.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell e.s.d.'s is used for estimating e.s.d.'s involving l.s. planes.

**Refinement.** Refinement of  $F^2$  against ALL reflections. The weighted  $R$ -factor  $wR$  and goodness of fit  $S$  are based on  $F^2$ , conventional  $R$ -factors  $R$  are based on  $F$ , with  $F$  set to zero for negative  $F^2$ . The threshold expression of  $F^2 > 2\sigma(F^2)$  is used only for calculat-

ing  $R$ -factors(gt) *etc.* and is not relevant to the choice of reflections for refinement.  $R$ -factors based on  $F^2$  are statistically about twice as large as those based on  $F$ , and  $R$ -factors based on ALL data will be even larger.

*Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters ( $\text{\AA}^2$ )*

	$x$	$y$	$z$	$U_{\text{iso}}^*/U_{\text{eq}}$
C1	0.8777 (2)	0.77555 (7)	0.9581 (2)	0.0472 (4)
C2	0.7685 (2)	0.77154 (8)	1.0017 (2)	0.0514 (5)
H2	0.7468	0.7983	1.0426	0.062*
C3	0.6814 (3)	0.72716 (8)	0.9882 (2)	0.0538 (5)
C4	0.5584 (2)	0.72918 (8)	1.0320 (2)	0.0514 (5)
C5	0.5266 (3)	0.76375 (9)	1.1109 (2)	0.0546 (5)
H5	0.5797	0.7924	1.1414	0.066*
C6	0.4014 (3)	0.74909 (11)	1.1383 (3)	0.0700 (7)
H6	0.3636	0.7670	1.1899	0.084*
C7	0.3450 (3)	0.70635 (11)	1.0805 (3)	0.0771 (8)
H7	0.2631	0.6918	1.0879	0.093*
C8	0.9616 (2)	0.82346 (8)	0.9849 (2)	0.0484 (5)
C9	0.8910 (2)	0.86269 (7)	0.8925 (2)	0.0455 (4)
C10	0.7624 (3)	0.86089 (9)	0.7771 (2)	0.0592 (6)
H10	0.7021	0.8336	0.7487	0.071*
C11	0.7326 (3)	0.90526 (11)	0.7067 (3)	0.0746 (8)
H11	0.6503	0.9106	0.6265	0.089*
C12	0.8367 (3)	0.93903 (10)	0.7685 (3)	0.0777 (8)
H12	0.8345	0.9702	0.7351	0.093*
C13	1.0565 (3)	0.76053 (11)	0.8186 (3)	0.0706 (7)
H13A	1.0041	0.7867	0.7605	0.106*
H13B	1.0911	0.7382	0.7698	0.106*
H13C	1.1402	0.7732	0.8937	0.106*
C14	0.8364 (2)	0.93124 (7)	0.28464 (19)	0.0406 (4)
C15	0.9289 (2)	0.91501 (8)	0.4059 (2)	0.0460 (4)
H15	0.8959	0.8903	0.4452	0.055*
C16	1.0792 (2)	0.93429 (7)	0.4794 (2)	0.0447 (4)
C17	1.1705 (2)	0.91180 (7)	0.6088 (2)	0.0441 (4)
C18	1.1315 (3)	0.87987 (9)	0.6871 (2)	0.0560 (5)
H18	1.0360	0.8676	0.6642	0.067*
C19	1.2562 (3)	0.86837 (11)	0.8071 (2)	0.0734 (7)
H19	1.2510	0.8478	0.8729	0.088*
C20	1.3836 (3)	0.89032 (10)	0.8162 (2)	0.0702 (7)
H20	1.4755	0.8861	0.8878	0.084*
C21	0.6899 (2)	0.90491 (7)	0.21273 (19)	0.0410 (4)
C22	0.5692 (2)	0.91621 (7)	0.25324 (18)	0.0402 (4)
C23	0.5668 (2)	0.94842 (8)	0.3484 (2)	0.0468 (4)
H23	0.6487	0.9667	0.4027	0.056*
C24	0.4256 (2)	0.95063 (9)	0.3547 (2)	0.0539 (5)
H24	0.4038	0.9705	0.4134	0.065*
C25	0.3249 (2)	0.92002 (9)	0.2642 (2)	0.0556 (5)
H25	0.2265	0.9168	0.2544	0.067*

## supplementary materials

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C26	0.7093 (3)	0.98422 (12)	0.0489 (3)	0.0780 (8)
H26A	0.6983	0.9558	-0.0052	0.117*
H26B	0.7164	1.0123	-0.0003	0.117*
H26C	0.6244	0.9873	0.0711	0.117*
O1	0.7101 (2)	0.68979 (6)	0.9424 (2)	0.0768 (5)
O2	1.08117 (19)	0.82733 (7)	1.07963 (18)	0.0719 (5)
O3	1.12945 (16)	0.96742 (6)	0.43439 (16)	0.0576 (4)
O4	0.67672 (18)	0.87642 (6)	0.12342 (16)	0.0575 (4)
S1	0.43580 (8)	0.68155 (2)	0.99352 (7)	0.0719 (2)
S2	0.97234 (7)	0.91905 (2)	0.91322 (7)	0.06458 (19)
S3	0.93242 (7)	0.72952 (2)	0.87829 (7)	0.06165 (17)
S4	1.35647 (6)	0.92628 (2)	0.68299 (6)	0.06146 (18)
S5	0.39848 (6)	0.88857 (2)	0.17198 (5)	0.05342 (16)
S6	0.87586 (6)	0.97902 (2)	0.20071 (6)	0.05825 (17)

### Atomic displacement parameters ( $\text{\AA}^2$ )

	$U^{11}$	$U^{22}$	$U^{33}$	$U^{12}$	$U^{13}$	$U^{23}$
C1	0.0487 (10)	0.0406 (10)	0.0460 (10)	-0.0015 (8)	0.0126 (8)	0.0032 (8)
C2	0.0557 (12)	0.0405 (11)	0.0569 (12)	-0.0059 (9)	0.0218 (10)	-0.0017 (9)
C3	0.0598 (12)	0.0433 (11)	0.0563 (12)	-0.0070 (9)	0.0212 (10)	0.0011 (9)
C4	0.0528 (11)	0.0419 (11)	0.0524 (11)	-0.0109 (8)	0.0141 (9)	0.0034 (9)
C5	0.0510 (11)	0.0528 (12)	0.0610 (13)	-0.0143 (9)	0.0235 (10)	-0.0053 (10)
C6	0.0630 (15)	0.0737 (17)	0.0787 (17)	-0.0072 (13)	0.0342 (13)	-0.0016 (14)
C7	0.0554 (14)	0.084 (2)	0.0864 (19)	-0.0167 (13)	0.0234 (13)	0.0172 (15)
C8	0.0455 (10)	0.0469 (11)	0.0490 (11)	-0.0066 (8)	0.0150 (9)	0.0013 (9)
C9	0.0414 (9)	0.0425 (10)	0.0521 (11)	-0.0059 (8)	0.0184 (8)	0.0000 (8)
C10	0.0500 (12)	0.0547 (13)	0.0604 (13)	-0.0071 (10)	0.0094 (10)	0.0107 (10)
C11	0.0472 (12)	0.0768 (17)	0.0822 (17)	-0.0016 (12)	0.0081 (12)	0.0322 (14)
C12	0.0544 (14)	0.0590 (15)	0.114 (2)	0.0032 (11)	0.0285 (14)	0.0331 (15)
C13	0.0684 (15)	0.0782 (18)	0.0718 (16)	-0.0030 (13)	0.0351 (13)	-0.0022 (13)
C14	0.0384 (9)	0.0375 (9)	0.0482 (10)	-0.0016 (7)	0.0198 (8)	-0.0032 (8)
C15	0.0439 (10)	0.0453 (11)	0.0489 (10)	-0.0059 (8)	0.0189 (8)	0.0008 (8)
C16	0.0397 (9)	0.0454 (11)	0.0492 (10)	0.0013 (8)	0.0183 (8)	-0.0006 (8)
C17	0.0414 (9)	0.0443 (10)	0.0451 (10)	-0.0011 (8)	0.0159 (8)	-0.0064 (8)
C18	0.0534 (12)	0.0670 (14)	0.0454 (11)	-0.0072 (10)	0.0178 (9)	0.0005 (10)
C19	0.0780 (17)	0.0847 (19)	0.0460 (12)	-0.0150 (14)	0.0132 (12)	0.0078 (12)
C20	0.0662 (15)	0.0754 (17)	0.0485 (12)	-0.0043 (13)	0.0020 (11)	-0.0006 (12)
C21	0.0413 (9)	0.0376 (9)	0.0421 (9)	-0.0026 (7)	0.0147 (7)	0.0006 (8)
C22	0.0371 (9)	0.0395 (9)	0.0399 (9)	-0.0050 (7)	0.0113 (7)	0.0010 (7)
C23	0.0453 (10)	0.0438 (11)	0.0484 (10)	-0.0015 (8)	0.0158 (8)	-0.0029 (8)
C24	0.0508 (11)	0.0574 (13)	0.0571 (12)	0.0043 (9)	0.0254 (10)	-0.0018 (10)
C25	0.0418 (10)	0.0674 (14)	0.0584 (12)	0.0021 (9)	0.0212 (9)	0.0077 (11)
C26	0.0602 (14)	0.0858 (19)	0.0701 (16)	-0.0064 (13)	0.0077 (12)	0.0305 (14)
O1	0.0985 (14)	0.0438 (9)	0.1046 (15)	-0.0114 (9)	0.0580 (12)	-0.0118 (9)
O2	0.0556 (9)	0.0694 (11)	0.0649 (10)	-0.0141 (8)	-0.0023 (8)	0.0120 (8)
O3	0.0433 (8)	0.0595 (9)	0.0660 (9)	-0.0073 (7)	0.0177 (7)	0.0125 (7)
O4	0.0603 (9)	0.0547 (9)	0.0599 (9)	-0.0101 (7)	0.0269 (7)	-0.0186 (7)

S1	0.0713 (4)	0.0563 (4)	0.0768 (4)	-0.0220 (3)	0.0182 (3)	-0.0003 (3)
S2	0.0508 (3)	0.0491 (3)	0.0863 (4)	-0.0106 (2)	0.0198 (3)	0.0032 (3)
S3	0.0664 (4)	0.0504 (3)	0.0718 (4)	-0.0007 (2)	0.0317 (3)	-0.0043 (3)
S4	0.0456 (3)	0.0647 (4)	0.0631 (3)	-0.0076 (2)	0.0105 (2)	0.0019 (3)
S5	0.0424 (3)	0.0632 (4)	0.0513 (3)	-0.0128 (2)	0.0155 (2)	-0.0078 (2)
S6	0.0483 (3)	0.0567 (3)	0.0638 (3)	-0.0093 (2)	0.0164 (2)	0.0152 (3)

*Geometric parameters (Å, °)*

C1—C2	1.339 (3)	C14—C15	1.340 (3)
C1—C8	1.521 (3)	C14—C21	1.514 (3)
C1—S3	1.740 (2)	C14—S6	1.733 (2)
C2—C3	1.465 (3)	C15—C16	1.463 (3)
C2—H2	0.9300	C15—H15	0.9300
C3—O1	1.225 (3)	C16—O3	1.230 (2)
C3—C4	1.465 (3)	C16—C17	1.467 (3)
C4—C5	1.400 (3)	C17—C18	1.382 (3)
C4—S1	1.714 (2)	C17—S4	1.716 (2)
C5—C6	1.433 (3)	C18—C19	1.417 (3)
C5—H5	0.9300	C18—H18	0.9300
C6—C7	1.345 (4)	C19—C20	1.355 (4)
C6—H6	0.9300	C19—H19	0.9300
C7—S1	1.683 (3)	C20—S4	1.686 (3)
C7—H7	0.9300	C20—H20	0.9300
C8—O2	1.213 (3)	C21—O4	1.215 (2)
C8—C9	1.448 (3)	C21—C22	1.453 (3)
C9—C10	1.375 (3)	C22—C23	1.372 (3)
C9—S2	1.719 (2)	C22—S5	1.7225 (18)
C10—C11	1.410 (3)	C23—C24	1.414 (3)
C10—H10	0.9300	C23—H23	0.9300
C11—C12	1.344 (4)	C24—C25	1.368 (3)
C11—H11	0.9300	C24—H24	0.9300
C12—S2	1.691 (3)	C25—S5	1.692 (2)
C12—H12	0.9300	C25—H25	0.9300
C13—S3	1.808 (3)	C26—S6	1.799 (3)
C13—H13A	0.9600	C26—H26A	0.9600
C13—H13B	0.9600	C26—H26B	0.9600
C13—H13C	0.9600	C26—H26C	0.9600
C2—C1—C8	116.85 (19)	C14—C15—C16	123.25 (18)
C2—C1—S3	124.61 (17)	C14—C15—H15	118.4
C8—C1—S3	118.52 (16)	C16—C15—H15	118.4
C1—C2—C3	123.4 (2)	O3—C16—C15	121.94 (18)
C1—C2—H2	118.3	O3—C16—C17	120.60 (18)
C3—C2—H2	118.3	C15—C16—C17	117.45 (18)
O1—C3—C2	121.6 (2)	C18—C17—C16	130.37 (19)
O1—C3—C4	121.2 (2)	C18—C17—S4	111.33 (16)
C2—C3—C4	117.3 (2)	C16—C17—S4	118.29 (15)
C5—C4—C3	129.90 (19)	C17—C18—C19	111.1 (2)
C5—C4—S1	111.32 (17)	C17—C18—H18	124.4

## supplementary materials

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C3—C4—S1	118.65 (17)	C19—C18—H18	124.4
C4—C5—C6	110.9 (2)	C20—C19—C18	113.2 (2)
C4—C5—H5	124.5	C20—C19—H19	123.4
C6—C5—H5	124.5	C18—C19—H19	123.4
C7—C6—C5	112.0 (3)	C19—C20—S4	112.3 (2)
C7—C6—H6	124.0	C19—C20—H20	123.9
C5—C6—H6	124.0	S4—C20—H20	123.9
C6—C7—S1	114.1 (2)	O4—C21—C22	122.99 (18)
C6—C7—H7	123.0	O4—C21—C14	119.72 (17)
S1—C7—H7	123.0	C22—C21—C14	117.27 (16)
O2—C8—C9	123.44 (19)	C23—C22—C21	129.21 (17)
O2—C8—C1	120.08 (19)	C23—C22—S5	111.39 (15)
C9—C8—C1	116.48 (17)	C21—C22—S5	119.35 (14)
C10—C9—C8	127.74 (19)	C22—C23—C24	112.33 (19)
C10—C9—S2	110.96 (16)	C22—C23—H23	123.8
C8—C9—S2	121.18 (15)	C24—C23—H23	123.8
C9—C10—C11	112.2 (2)	C25—C24—C23	112.0 (2)
C9—C10—H10	123.9	C25—C24—H24	124.0
C11—C10—H10	123.9	C23—C24—H24	124.0
C12—C11—C10	112.4 (2)	C24—C25—S5	112.74 (17)
C12—C11—H11	123.8	C24—C25—H25	123.6
C10—C11—H11	123.8	S5—C25—H25	123.6
C11—C12—S2	113.1 (2)	S6—C26—H26A	109.5
C11—C12—H12	123.4	S6—C26—H26B	109.5
S2—C12—H12	123.4	H26A—C26—H26B	109.5
S3—C13—H13A	109.5	S6—C26—H26C	109.5
S3—C13—H13B	109.5	H26A—C26—H26C	109.5
H13A—C13—H13B	109.5	H26B—C26—H26C	109.5
S3—C13—H13C	109.5	C7—S1—C4	91.68 (13)
H13A—C13—H13C	109.5	C12—S2—C9	91.37 (12)
H13B—C13—H13C	109.5	C1—S3—C13	102.81 (12)
C15—C14—C21	118.14 (17)	C20—S4—C17	92.08 (12)
C15—C14—S6	124.49 (15)	C25—S5—C22	91.55 (10)
C21—C14—S6	117.31 (14)	C14—S6—C26	103.40 (11)
C8—C1—C2—C3	-177.6 (2)	C16—C17—C18—C19	-178.5 (2)
S3—C1—C2—C3	0.5 (3)	S4—C17—C18—C19	0.4 (3)
C1—C2—C3—O1	4.3 (4)	C17—C18—C19—C20	-1.1 (4)
C1—C2—C3—C4	-175.8 (2)	C18—C19—C20—S4	1.3 (4)
O1—C3—C4—C5	165.1 (2)	C15—C14—C21—O4	-101.2 (2)
C2—C3—C4—C5	-14.8 (4)	S6—C14—C21—O4	76.3 (2)
O1—C3—C4—S1	-10.4 (3)	C15—C14—C21—C22	80.2 (2)
C2—C3—C4—S1	169.69 (17)	S6—C14—C21—C22	-102.33 (18)
C3—C4—C5—C6	-175.1 (2)	O4—C21—C22—C23	-177.7 (2)
S1—C4—C5—C6	0.7 (3)	C14—C21—C22—C23	0.8 (3)
C4—C5—C6—C7	-0.7 (3)	O4—C21—C22—S5	-0.2 (3)
C5—C6—C7—S1	0.4 (3)	C14—C21—C22—S5	178.31 (14)
C2—C1—C8—O2	98.6 (3)	C21—C22—C23—C24	177.5 (2)
S3—C1—C8—O2	-79.6 (3)	S5—C22—C23—C24	-0.2 (2)
C2—C1—C8—C9	-81.4 (3)	C22—C23—C24—C25	0.0 (3)



S3—C1—C8—C9	100.4 (2)	C23—C24—C25—S5	0.1 (3)
O2—C8—C9—C10	174.5 (2)	C6—C7—S1—C4	0.0 (2)
C1—C8—C9—C10	-5.5 (3)	C5—C4—S1—C7	-0.39 (19)
O2—C8—C9—S2	-1.2 (3)	C3—C4—S1—C7	175.9 (2)
C1—C8—C9—S2	178.76 (15)	C11—C12—S2—C9	0.7 (3)
C8—C9—C10—C11	-175.6 (2)	C10—C9—S2—C12	-0.7 (2)
S2—C9—C10—C11	0.5 (3)	C8—C9—S2—C12	175.7 (2)
C9—C10—C11—C12	0.1 (4)	C2—C1—S3—C13	169.9 (2)
C10—C11—C12—S2	-0.6 (4)	C8—C1—S3—C13	-12.07 (19)
C21—C14—C15—C16	173.87 (18)	C19—C20—S4—C17	-0.9 (2)
S6—C14—C15—C16	-3.4 (3)	C18—C17—S4—C20	0.30 (19)
C14—C15—C16—O3	0.9 (3)	C16—C17—S4—C20	179.33 (18)
C14—C15—C16—C17	-177.61 (19)	C24—C25—S5—C22	-0.19 (19)
O3—C16—C17—C18	169.2 (2)	C23—C22—S5—C25	0.21 (17)
C15—C16—C17—C18	-12.3 (3)	C21—C22—S5—C25	-177.70 (17)
O3—C16—C17—S4	-9.7 (3)	C15—C14—S6—C26	-178.4 (2)
C15—C16—C17—S4	168.90 (15)	C21—C14—S6—C26	4.32 (19)

Hydrogen-bond geometry ( $\text{\AA}$ ,  $^\circ$ )

$D-H\cdots A$	$D-H$	$H\cdots A$	$D\cdots A$	$D-H\cdots A$
C2—H2 $\cdots$ O4 <sup>i</sup>	0.93	2.52	3.448 (3)	172
C5—H5 $\cdots$ O4 <sup>i</sup>	0.93	2.54	3.416 (3)	157
C20—H20 $\cdots$ O4 <sup>ii</sup>	0.93	2.54	3.438 (3)	162

Symmetry codes: (i)  $x, y, z+1$ ; (ii)  $x+1, y, z+1$ .

Fig. 1

